

Original Research

Environmental Occurrence and Ecological Risk of Bisphenol A in Erhai Lake Basin Away From Industrial Regions in China

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Abstract

The environmental distribution and migration of bisphenol A (BPA) has attracted attention due to its widespread use. The Erhai Lake Basin is dominated by agricultural production and tourism, and almost no industry. In this research, BPA pollution in the non-industrial region was investigated. BPA was found in all samples of inflow rivers, with concentrations ranging 2.80 ng/L~61.85 ng/L. The measured 15 rivers that account for 78.7% of the total inflow volume were inputting BPA concentration of 20.69 ng/L into Erhai Lake with an annual load of 8.54 kg/year. The measured BPA removal efficiency of sewage treatment plants ranged 49.8%~92.8%. Oxidation Ditch and A²O-MBR treatment process achieved the better BPA removal efficiency. The amount of BPA discharged directly to Erhai Lake from the effluent of the sewage plants each year was 1.63 kg. Based on the above research, a multimedia model of QWASI was established, and water concentration calculated was 0.203ng/L. A PNEC value considering more sensitive receptors was applied to this study, and it was indicated that there was a certain ecological risk in the water of the Boluo River estuary, while other rivers and lake were relatively safe. It was revealed that the ecological risk of BPA was relatively low in the Erhai Lake.

Keywords: BPA, non-industrial regions, occurrence, ecological risk, multimedia model

Introduction

Bisphenol A, which is also called BPA, has been predominantly used in many consumer products, including plastics (as polymers, polycarbonate plastics), PVC, food packaging, dental sealants, and thermal receipts. As one of the most produced and consumed

chemicals in the world, BPA has become a public health problem through extensive and continuous exposure [1, 2]. Many scholars paid attention to the exposure and release of BPA. A related study in Taiwan [3] pointed out that there was almost no risk of BPA intake through diet, and BPA was ingested by the human body mainly due to other exposures. Morgan [4] studied the content of BPA in the urine of 50 American adults and found that the median BPA concentration in human urine was 2.07 µg/L. Through their life contact observation,

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the proportion of BPA intake from the diet was about 20%, while contacting other sources seemly dominate. Woerden [5], Karrer [6] and many scholars also obtained similar research conclusions [7]. It has been proved that BPA exists mainly in natural waters, sediments, soil, as well as sewage and wastewater sludge, with a concentration ranging from nanogram to microgram level [8-10], posing a greater threat to humans and animals. After entering human body, BPA can disrupt normal cellular functions by acting as an estrogen agonist and an androgen antagonist [11]. It is suspected that BPA may affect human development throughout the fetal period [12] and may cause cancer. Because of its estrogen activity, it has been shown to reduce sperm count and sperm activity, and to be toxic to the liver, and may even be associated with chronic human diseases such as cardiovascular disease and diabetes.

BPA could migrate and transform in environment through processes, including diffusion, adsorption, and degradation. It is difficult to hydrolyze under natural conditions. The response strategies to control BPA pollution are mainly through various methods to decompose, transform and remove. Currently, the research methods of BPA degradation include physical and chemical methods, microbial degradation, and advanced oxidation [13, 14]. Bertanza's research [9] indicated that using different wastewater treatment process schemes, such as MBR (membrane bioreactor) and CASS (circulating activated sludge system), the BPA removal rate can reach 70%. Hu's study [15] clarified the conversion products of BPA in the ozonation

process of sewage plants, aerobic soils and water. Soil is also one of the important media for BPA adsorption and degradation, and BPA has a fast mineralization rate and poor migration in soil [16]. Chen [17] found that sodium alginate/ultrasound assistance could significantly improve the biodegradation rates of BPA in soil.

Based on the characteristics of BPA industrial sources, people usually think that BPA pollution in industrial areas is more serious, while BPA in remote, non-industrial areas, or agricultural economically dominant areas will be lower. In this research, BPA pollution in the non-industrial region was investigated. The study also analyzed the removal efficiency of BPA by major sewage treatment plants. Based on this, a fugacity multimedia model was established to simulate the BPA occurrence status in various environmental media and evaluate environmental risk in the Erhai Lake.

Materials and Methods

Overview of Study Area

Erhai Lake is located in Yunnan Province, Southwest of China, and it is an important source of drinking water in Dali Prefecture, Yunnan Province. The basin has a total area of 2565 km², including lake area of 252.1 km². Erhai Lake has a maximum water depth of 21.3 m, an average water depth of 10.8m, and a lake capacity of 2.959 billion m³. Erhai Lake is a typical plateau lake with a small annual runoff. The lake



Fig. 1. Study area location and sampling sections.

water exchange cycle takes several years. Due to its remote location, the Erhai Lake Basin is dominated by agricultural production and tourism, and there is almost no industry. Therefore, the water quality of Erhai Lake has been excellent so far.

Site Selection

Erhai Lake has 117 inflow rivers with yearly average inflow of 825 million m³, and only one outflow river, which is called Xier River (upper Mekong River). The three rivers with the largest inflow are located in the north of Erhai Lake, namely Luoshi River, Yongan River and Miju River, contributing 50% of total inflow

volume. Other large inflow rivers also include the Boluo River in the south and 18 streams in the west. This study selected the 15 largest rivers flowing into Erhai Lake, setting up sampling sections for BPA detection at the estuary. The sampling sections location are shown in Fig. 1. In addition, in order to investigate the BPA removal efficiency of sewage treatment plants in Erhai Basin. The inlet and outlet water of the main sewage treatment plants were sampled to detect the BPA concentration. Information of sampling sections as Table 1. In the preliminary work, we regularly monitored the flow volume of selected 15 rivers into the lake, and the data are also listed in Table 1. Two samples were taken at each sampling section.

Table 1. Information of sampling sections.

Sampling number	Name of infow river	Latitude and longitude of estuary	Annual inflow volume -2019 (×10 ⁴ m ³)
S1	Luoshi River	25.943508°N, 100.098616°E	6609.807
S2	Miju River	25.926798°N, 100.133399°E	9017.867
S3	Yongan River	25.954445°N, 100.141001°E	5615.688
S4	Wanhua River	25.868505°N, 100.140747°E	683.209
S5	Yang River	25.819498°N, 100.143812°E	2483.872
S6	Mangyong River	25.799785°N, 100.143074°E	3478.54
S7	Jin River	25.770484°N, 100.148407°E	694.553
S8	Mei River	25.736493°N, 100.174900°E	88.757
S9	Tao River	25.733980°N, 100.178460°E	216.5
S10	Zhonghe River	25.718848°N, 100.187559°E	122.931
S11	Baihe River	25.699636°N, 100.204390°E	267.355
S12	Long River	25.686205°N, 100.205739°E	654.061
S13	Qingbi River	25.678176°N, 100.209512°E	359.232
S14	Mocan River	25.673353°N, 100.207481°E	442.71
S15	Boluo River	25.614500°N, 100.281889°E	1739.246
Sum of annual flow above			32474.328
All inflow to the lake in 2019			41257.69
Sampling number	Major sewage treatment plants	latitude and longitude of plants	Inlet or outlet
D1i	Dayutian	25.574493°N, 100.199249°E	Inlet
D1o			Oulet
D2i	Haidong	25.699396°N, 100.274867°E	Inlet
D2o			Oulet
D3i	Shuanglang	25.922345°N, 100.192492°E	Inlet
D3o			Oulet
D4i	Shangguan	25.955189°N, 100.130676°E	Inlet
D4o			Oulet
D5i	Zhoucheng	25.901317°N, 100.109351°E	Inlet
D5o			Oulet

Sampling Procedure

Samples collection of surface water was under the water surface 5cm in order to prevent floating objects, suspended oil and other impurities on the water surface from mixing into the sampler. Water samples of urban sewage treatment plants were taken from the sewage inlet and the total discharge port of the sewage treatment facility. The glass container required for sampling should be cleaned in advance. The sampling position, sampling time and sampling temperature were recorded during material collection. The water samples taken by the sampler were transferred to a brown glass bottle with a capacity of 1L filled completely, sealed and taken back to the laboratory to be refrigerated in a 4°C refrigerator. Considering that other microbial reactions could cause analysis errors, water samples were fixed with 1 g/L ascorbic acid. In order to ensure the accuracy of the analysis results, the samples were processed by solid phase extraction within one week and sent to the GC-MS for testing within two weeks.

SPE Extraction and Derivatization

Before extraction of the sample, the solid-phase extraction cartridge was rinsed with a suitable solvent, so that the analyte could be in close contact with the surface of the solid phase, thereby improving the recovery rate and reproducibility. In this experiment, 2×3 mL of n-hexane, 2×3 mL of ethyl acetate, 2×3 mL of methanol, and 2×5 mL of ultrapure water were used to activate the column. The dripping speed of the liquid was about 1mL/min and time 5 minutes to make sure that the filler was thoroughly soaked in the activating reagent. 0.5~1.0 L of pre-treated water sample was imported into the activated SPE cartridges under a certain vacuum pressure, while flow rate was controlled at about 4 mL/min, then the BPA and some impurities were adsorbed and retained on the SPE cartridges. The cartridges were washed by 10 mL of 10% methanol in water with flow rate of approximately 5 mL/min. The cartridges were eluted with 10 mL of ethyl acetate, while the drip rate was controlled at about 1 mL/min, and the eluate was collected. Dried BPA eluent under mild high-purity nitrogen at 30°C was added 25 µL of BSTFA (containing 1% by mass TMCS) reagent and 50 µL of pyridine solution, derivatized with microwave 300 W for 4 minutes. After nitrogen blowing, 400 µL of 1 mg/L internal standard Mirex was added to dissolve BPA, and then transferred to a 2 mL sample bottle for detection.

Analysis by GC-MS

The testing instrument was Thermo Fisher TRACE 1300-ISQ Series Quadrupole. GC conditions: TR-5MS quartz capillary column (30 m×0.25 mm×0.25 µm); high-purity helium (99.999%) carrier gas, constant current mode (1.2 mL/min carrier gas flow rate);

splitless injection, each sampling volume was 1.0 µL, with the inlet temperature of 280°C; septum purge 5.0 mL/min; column heating program: initial temperature 50°C, hold for 2 min; rise to 260°C at the rate of 20°C/min, hold for 5 min; then increase to 280°C at the rate of 10°C/min, holding for 5 minutes. MS condition: Electron bombardment ion source (EI), ionization voltage 70eV; ion source temperature 250°C, transmission line temperature 280°C; solvent delay time 10min; full scan mode (SCAN) for qualitative and selective ion scan mode (SIM) for quantification. The analysis was performed using Xcalibur software to determine BPA and internal standard.

In order to examine the accuracy and precision of the method, a standard solution containing 50ng and 500ng of the test component was used as a sample, and the measurement was performed 7 times within one day. The average spike recovery of BPA was 98.0%. The relative standard deviation (RSD) was 0.4%~6.8%. Limits of quantification were defined as 10 times the noise level of the baseline in the chromatograms. For data analysis, all data of result shown through in figures are expressed as mean±standard deviation.

Multimedia Model Calculation

Currently, widely used multi-media model is the fugacity model proposed by Professor Donald Mackay of the Environmental Model Research Center of the University of Trent, Canada. This method is based on the fugacity concept and uses the principle of mass balance. Researchers have proposed a number of different multimedia fugacity models. These models have been softwareized and provided to users for related scientific research, including SimpleBox, QWASI, CalTOX, ChemCAN, EQC, TaPL3, etc., while each of these models are not very different, slightly different according to various objects. For example, QWASI is a simplified lake model. The purpose is to analyze and calculate the chemical exchange of chemicals in water, air and water-sediment and the chemical transformation in a single phase, so it is not suitable for chemical behavior analysis of land and rivers; EQC sets up a fixed multimedia environment system, which cannot be directly used for regional environment simulation; SimpleBox, CalTOX, ChemCAN, and TaPL3 can be applied to chemicals in regions. On the distribution and risk assessment of environmental media, the reliability of the models have been verified [18-27]. The fugacity model also has obvious shortcomings, because the existing fugacity models assume that the spatial characteristics of each medium are uniform, that is, the fugacity of each point in each medium is equal. If the model is used for heterogeneous phases, the behavior characteristics of pollutants cannot be reflected spatial variability, the accuracy of the model would be reduced. Taking the Erhai Lake Basin as an example. Erhai Lake is a typical plateau lake, showing a small annual runoff. Dozens of rivers flow into the lake but only

one river out. Therefore, lake's water body stays for several years, and the detention time of inflow rivers sub-basin is short but the detention time in the lake is very long. Based on the environmental characteristics of the Erhai Lake Basin, in order to use the multimedia model to study the environmental behavior of BPA, in this study, monitoring of rivers and sewage plants were implemented. After all, human production and living behaviors occur on land and near rivers. For the Erhai Lake Basin, there is another important factor. At present, local governments have built a pollution interception system around the lake. Except for rivers and effluent of sewage plants that mainly flow into the lake, there are basically no cases where pollutants directly enter the lake.

In this study, the QWASI model (version 3.10) was employed, which is a Mackay-type multimedia fate model describing the fate of a chemical in a given lake. The QWASI model is suitable for water bodies with uniformly mixed water and particle streams. When the system is in a steady state, that is, the lake is exposed to constant input conditions for a long time, the sum of all input rates is equal to the sum of all output rates. Based on the diversification of BPA exposure pathways, it is difficult to estimate the production and emissions of BPA and downstream products within the Erhai Lake Basin. In this study, the pathway of BPA release into the environment is assumed to be through sewage discharge. The amount of emissions was calculated based on the removal efficiency of sewage treatment plants, which was tested in this study. BPA concentration data input into Erhai Lake based on experimentally measured river data. Experimentally measured data were used as BPA concentration of advection into Erhai Lake in the QWASI.

The required data was input to apply the QWASI model. The physicochemical properties of BPA and the geographical parameters for Erhai Lake are summarized in Table 2.

Results and Discussion

Distribution of BPA in Inflow Rivers of Erhai Lake

Fig. 2 shows distribution of BPA in inflow rivers of Erhai Lake. BPA was found in all samples, with concentrations ranging from 2.80 ng/L to 61.85 ng/L. Concentration levels were similar to the studies by Moreira [28], Arditoglou [29], Funakoshi [30], Fu [31], Kuch [32] and Loos [33], showing BPA occurrence of rivers and estuaries in these locations including Netherlands, Brazil, Qingdao of China, Japan, Germany and Belgium. In the Erhai Lake Basin, the concentration of the three rivers (Luoshi River, Miju River and Yongan River) with the largest inflow to the lake was between 13.4 ng/L and 24.6 ng/L, which represented the main pollutant transport channel of the river into the lake. The river with the highest concentration was the Boluo River. Boluo River sub-basin was where the Dali Economic Development Zone was located. The vast majority of industrial enterprises in the Erhai Lake Basin were located here, including tractor manufacturers, organic fertilizer production plants, livestock and poultry farms, transportation and logistics bases, and automobile inspection and trading markets. Industry, commerce, and a large number of production and living activities might contribute a lot of BPA emission. However, compared with the BPA river water concentration of 5.03 $\mu\text{g/L}$ ~12 $\mu\text{g/L}$ in some areas of Portugal or the United States [34], the BPA pollution in the inflow rivers of Erhai Lake Basin was much less important.

BPA Removal Efficiency by Sewage Treatment Plants

Dayutian Sewage Plant is the most important sewage treatment plant in Dali City, with treatment capacity of

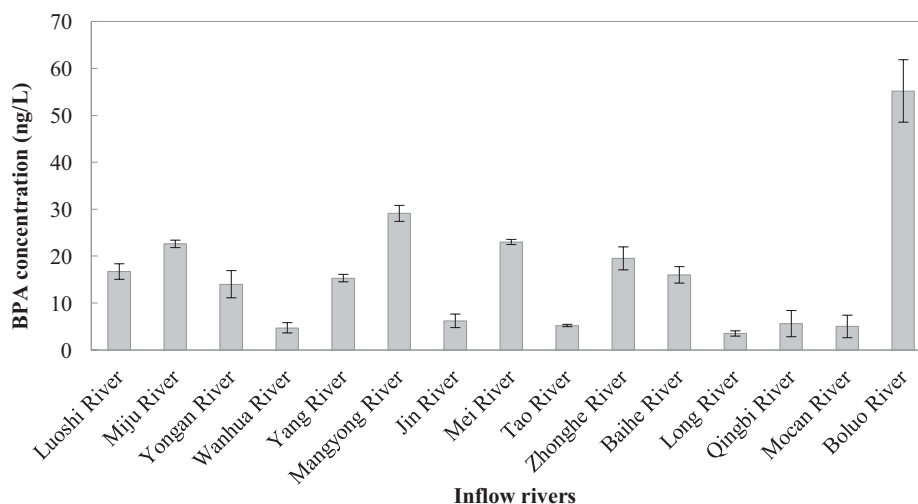


Fig. 2. Distribution of BPA in inflow rivers of Erhai Lake.

Table 2. Geographic parameters and chemical parameters.

Parameter	Unit	Parameter value
BPA		
Molar mass	g/mol	228
Temperature	°C	25
Vapour pressure	Pa	4.0E-06
Water solubility	g/m ³	300
LogKow	/	2.76
Melting point	°C	159
Degradation half-life in water ^a	h	363.2
Degradation half-life in sediment ^a	h	3007.8
Lake properties		
Water surface area	m ²	252.1E+06
Water volume	m ³	29.59E+08
Sediment active layer depth	m	0.5
Concentration of solids in water column	mg/L	2
Concentration of solids in inflow water	mg/L	5
Concentration of solids in sediment	m ³ /m ³	0.3
Density of solids in water	kg/m ³	2400
Density of solids in sediment	kg/m ³	2400
Density of solids in aerosols	kg/m ³	1500
Organic carbon fraction of solids in water column	g/g	0.2
Organic carbon fraction of solids in sediment	g/g	0.04
Organic carbon fraction of solids in inflow water	g/g	0.2
Organic carbon fraction of solids in resuspended sediment	g/g	0.04
Flows and transfer coefficients		
River water inflow ^b	m ³ /h	47097.82
Water outflow rate ^b	m ³ /h	55365.3
Deposition rate of solids	g/(m ³ ·d)	1.4
Burial rate of solids	g/(m ³ ·d)	0.5
Resuspended rate of solids	g/(m ³ ·d)	0.5
Aerosol dry deposition velocity	m/h	10.8
Scavenging ratio	vol air per vol rain	2.0E+05
Rain rate	m/year	0.82
Air side mass transfer coefficients of volatilization	m/h	3
Water side transfer coefficients of volatilization	m/h	0.03
Mass transfer coefficients of sediment-water diffusion	m/h	1.0E-04
Emissions and inflows		
Direct discharges into water ^c	kg/year	1.63
Chemical concentration in inflow water ^c	ng/L	20.69

a: Reference to chemical database of the SimpleBox 4.0 model (developed by the Dutch Institute of Public Health and the Environment). b: Data collected in this study. c: Data through measured and calculated at each of the studied rivers and plants.

129,000 m³/d, and it has accepted the main domestic sewage in Dali City. The other sewage treatment plants mainly treat township domestic sewage, with the treatment capacity of 5,000 m³/d (Haidong Plant), 2,000 m³/d (Shuanglang Plant), 2,000 m³/d (Shangguan Plant) and 2,000 m³/d (Zhoucheng Plant). By measuring the BPA inlet and outlet of the sewage plants (shown in Fig. 3), we found that the sewage treatment plants of Erhai Lake Basin had insufficient treatment efficiency for BPA. The measured BPA removal efficiency ranged from 49.8%~92.8%. The sewage treatment plant with the highest treatment efficiency was Zhoucheng Plant, which also had the highest intake BPA concentration of 563.7 ng/L. The high inlet concentration might be due to the distribution of some ethnic dyeing factories near Zhoucheng. Data of Dayutian Sewage Plant could reflect the BPA release concentration (302.3 ng/L) of Dali urban residents. Woerden's research [5] indicated that average BPA concentration of human urinary was 2.3 μg/L. Considering the dilution effect of other types of sewage, the measured concentration levels were reasonable in the absence of industrial emissions. In addition, the treatment process of sewage plants also had a significant impact on BPA removal efficiency. A²O-MBR process was used in Dayutian Plant, and Orbel Oxidation Ditch in Zhoucheng Plant. Due to lower cost, other township sewage treatment plants (Haidong Shuanglang and Shangguan) applied diatomite flocculation sedimentation process. Viewing from BPA removal efficiency, Oxidation Ditch>A²O-MBR>Flocculation.

Occurrence and Ecological Risk of BPA in Erhai Basin

Assuming that the BPA advection from the air into the Erhai Lake was ignored, the measured BPA concentration and flow volume data of all inflow rivers

were applied to calculate the concentration of BPA input through advection of river water, calculating as follows.

$$C_{BPA} = \frac{\sum_i^n (C_i \times AF_i)}{\sum_i^n AF_i} \quad (1)$$

C_{BPA} : average BPA concentration in inflow rivers; n : number of BPA measured inflow rivers; i : a measured inflow river; C_i : BPA concentration in a inflow river; AF_i : annual flow volume of a measured inflow river.

Based on the excellent pollution interception system of Erhai Lake, it was assumed that the BPA discharge channel was through only sewage treatment plants outlet. Mass of BPA emissions to Erhai Lake was calculated as follows.

$$E_{BPA} = \sum_j^m C_j \times V_j \times N \quad (2)$$

E_{BPA} : Mass of BPA directly discharged to Erhai Lake per year; m : number of sewage treatment plants; j : a measured sewage treatment plant; C_j : BPA concentration in effluent of a sewage treatment plant; V_j : treated water volume of a sewage treatment plant per day; N : days per year, value was 365.

Through the established QWASI model and related input parameters, occurrence and fate of BPA in Erhai Lake was simulated. The predicted environmental concentrations in the four bulk environmental compartments of air, water and sediment were calculated. Mass balances and diagram of QWASI model output results are shown in Table 3 and Fig 4. It can be clearly seen that bulk water account for 96.71% of total mass in the system. Water and sediment concentrations calculated in QWASI were 0.203 ng/L and 2.04E-04 ng/g, respectively. The output showed

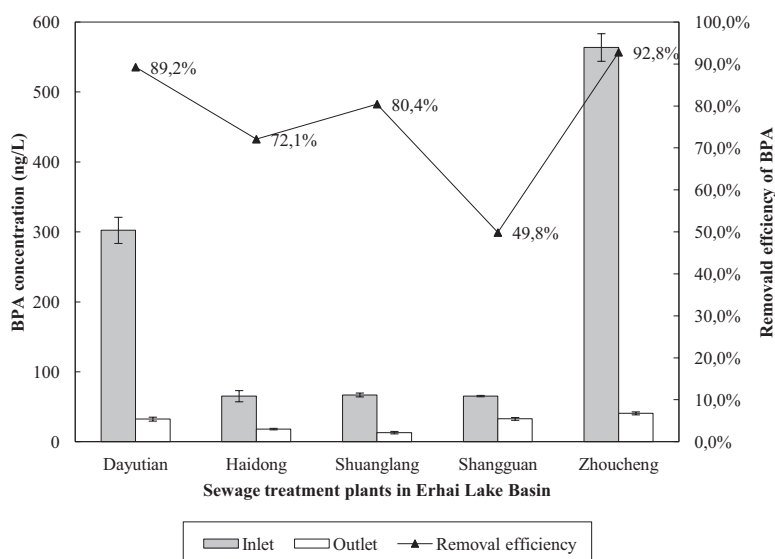


Fig. 3. BPA removal efficiency sewage treatment plants of Erhai Lake Basin.

Table 3. Mass balances in QWASI simulation.

In the System	Unit (mol/h)	Unit (kg/year)
Total chemical inputs	5.09E-03	10.17
Emission	8.16E-04	1.63
Inflow	4.27E-03	8.54
Total chemical losses	5.09E-03	10.17
Outflow	4.92E-05	0.0983
Water to air transfer	8.25E-10	1.65E-06
Total transformation	5.04E-03	10.07
Sediment burial	4.70E-09	9.38E-06

that BPA pollution in Erhai Lake was at a relatively low level.

Chronic predicted no effect concentrations (PNECs) for aquatic systems are used to support the assessment of potential risks to aquatic organisms in receiving waters. PNECs for a compound are considered protective of populations, communities, and ecosystems. Traditionally, PNECs are derived by taking the lowest no-observed effect concentration (NOEC) from a set of toxicity studies and dividing by an assessment factor [35]. Species Sensitivity Distribution (SSD methods) was applied by Staples and a PNEC value of 1.6 $\mu\text{g/L}$ were derived. Adamakis [36] investigated the effects of environmentally relevant BPA concentrations (0.03~3 $\mu\text{g/L}$) on leaf elongation and cytoskeleton in the seagrass *Cymodocea nodosa*. They found that the highest BPA concentration (3 $\mu\text{g/L}$)

significantly affected leaf elongation at the beginning of the experiment, and cytoskeleton disturbance was observed even at lower concentrations (around 0.03 $\mu\text{g/L}$). Wright-Walters [37] thought that a BPA PNEC of 0.06 $\mu\text{g/L}$ was more appropriate. Because aquatic receptors were more sensitive to BPA, their exposure was greater than previously found. This was due to the fact that Wright-Walters' study had higher sensitivity taxa, such as algae, cyanobacteria, and insects. In ecological assessment, it was assumed that the sensitivity of the ecosystem (PNEC) depended on the most sensitive species. With this PNEC, ecosystems including insects, molluscs, and crustaceans could be protected from the adverse effects of exposure to BPA.

Combining the PNEC value with the data in our study, we found that there was a certain ecological risk in the water of the Boluo River estuary, other rivers were relatively safe, and the effluent concentration of the sewage plant was lower than the PNEC value, which was worth to be pleased with. However, in recent years, with the increase of China's polycarbonate production capacity and output, and the steady growth of the epoxy resin market, China's BPA demand has generally shown a relatively stable growth trend. In 2018, China's apparent consumption of BPA reached 16.675 million tons. Due to wide variety of products and applications, a total production capacity of BPA exceeded 3.8 million tons [38], and apparent consumption of BPA is becoming more and more in China, which is an important BPA producer (Fig. 5). Based on the diversity of BPA products and exposure routes, we should continue to pay attention to the environmental distribution of BPA.

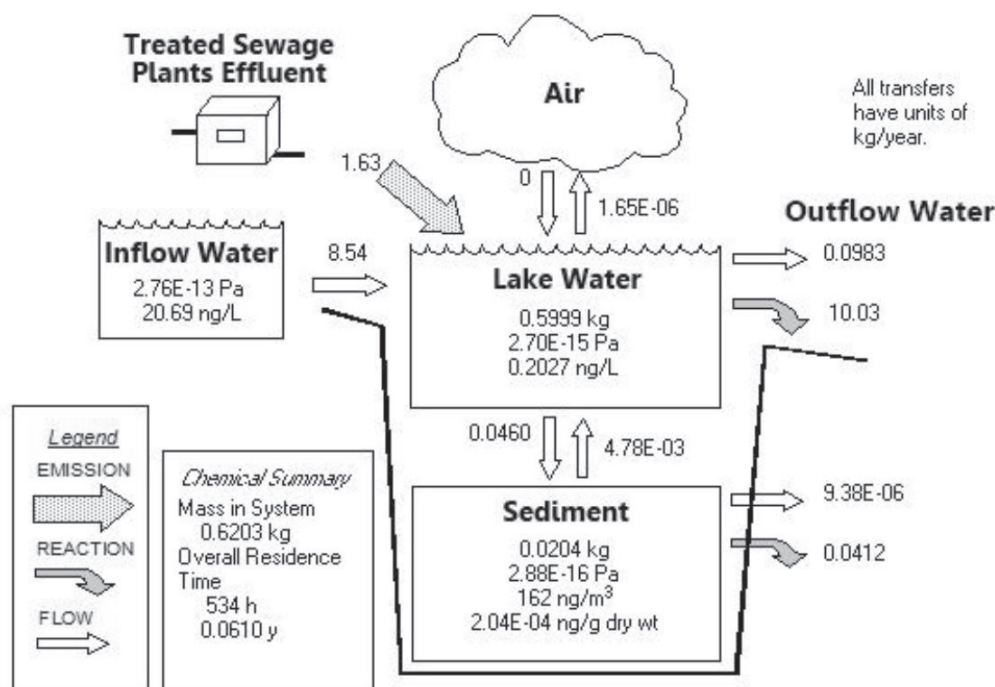


Fig. 4. Fate of BPA in Erhai Lake.

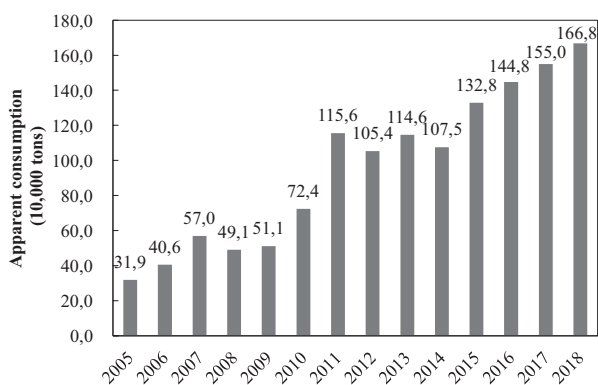


Fig. 5. China's apparent consumption of BPA.

Conclusions

Erhai Lake is a typical plateau lake in China. The focus on BPA prompted us to conduct this study due to the long residence time of chemicals in the lake.

The results of this research explained that BPA was found in all samples of inflow rivers, with concentrations ranging from 2.80 ng/L to 61.85 ng/L. The measured 15 rivers that account for 78.7% of the total inflow volume were inputting BPA concentration of 20.69 ng/L into Erhai Lake with an annual load of 8.54 kg/year.

By measuring the BPA inlet and outlet of the sewage plants, we found that the sewage treatment plants of Erhai Lake Basin had insufficient treatment efficiency for BPA. The measured BPA removal efficiency ranged 49.8%–92.8%. Oxidation Ditch and A²O-MBR treatment process achieved the better BPA removal efficiency. The amount of BPA discharged directly to Erhai Lake from the effluent of the sewage plants was 1.63 kg/year.

Through the established QWASI model and related input parameters, occurrence and fate of BPA in Erhai Lake was simulated. Water concentration calculated was 0.203 ng/L, showing that BPA pollution in Erhai Lake was at a relatively low level. A PNEC value considering more sensitive receptors was applied to this study, and it was indicated that there was a certain ecological risk in the water of the Boluo River estuary, while other rivers and lake were relatively safe. It was revealed that the ecological risk of BPA was relatively low in the Erhai Lake Basin, which is far from the heavy and chemical industrial areas. Based on the diversity of BPA products and exposure routes, the environmental distribution and the ecological risk of BPA should be paid more attention to.

Acknowledgements

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Conflict of Interest

The Authors declare no conflict of interest.

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